

This article was downloaded by:

On: 25 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Liquid Crystals

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713926090>

### Characterization of the B<sub>4</sub> phase by dielectric and AFM measurements

H. Kresse; J. Salfetnikova; H. Nadasi; W. Weissflog; A. Hauser

Online publication date: 06 August 2010

**To cite this Article** Kresse, H. , Salfetnikova, J. , Nadasi, H. , Weissflog, W. and Hauser, A.(2011) 'Characterization of the B<sub>4</sub> phase by dielectric and AFM measurements', *Liquid Crystals*, 28: 7, 1017 – 1023

**To link to this Article:** DOI: 10.1080/02678290110039093

**URL:** <http://dx.doi.org/10.1080/02678290110039093>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

# Characterization of the B<sub>4</sub> phase by dielectric and AFM measurements

H. KRESSE\*, J. SALFETNIKOVA, H. NADASI, W. WEISSFLOG  
and A. HAUSER

Martin-Luther-Universität Halle-Wittenberg, Institut für Physikalische Chemie,  
Mühlpforte 1, D-06108 Halle, Germany

(Received 18 September 2000; accepted 11 December 2000)

Dielectric measurements on a sample consisting of banana-shaped molecules were carried out in a frequency range between 0.1 Hz and 10 MHz. The sample exhibited B<sub>2</sub> and B<sub>4</sub> phases. As usual, two ranges of relaxation were detected in the B<sub>2</sub> phase, the fast reorientation about the long axes of the molecules and a slow collective process. In the B<sub>4</sub> phase, only one dielectric active process at low frequencies was found; this does not differ from the low frequency relaxation of the B<sub>2</sub> modification. This relaxation is probably related to the dynamics of superstructures. Crystallization could be observed after keeping the sample for a longer time at higher temperature. Thus, it was possible to differentiate clearly between the crystalline and the B<sub>4</sub> phases. AFM investigations prove the existence of focal-conic domains and periodic superstructures in the B<sub>4</sub> phase; then do not appear in the crystalline state. For this reason the B<sub>4</sub> phase is regarded as different from a classical crystalline phase.

## 1. Introduction

In 1996 Niori *et al.* [1] discovered the first members of a new system of mesomorphic phases formed by banana-shaped molecules and situated between the isotropic and solid states. To date, eight polymorphic phases are known and most of them have been characterized by X-ray measurements [2–4]. Link *et al.* [5] and Brand *et al.* [6] have demonstrated that there are different possibilities for the arrangement of such molecules, and so different superstructures can be formed. Dielectric investigations can help to understand the specific relations in phases resulting from bent-shaped molecules, which are in part comparable with the different modifications of the chiral smectic C phase (see for example [7]).

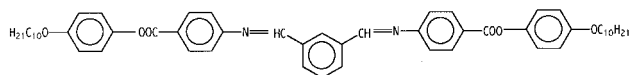
In previous investigations it was demonstrated that the B<sub>1</sub> [8], B<sub>2</sub> [9–11], B<sub>3</sub> [12], B<sub>7</sub> and B<sub>8</sub> [13] phases show typical dielectric behaviours which can be partially used as a ‘fingerprint’ for the different modifications. The B<sub>4</sub> phase was first investigated by the dielectric method in [14] and later in [15]. In the first publication, a low frequency absorption was expected, but could not clearly be proven. The existence of this relaxation was then demonstrated for a highly purified sample in [15] by heating and cooling. The sample involved did not crystallize, and so it was difficult to differentiate between a solid and the B<sub>4</sub> phase. In this paper, a sample is

considered which exhibits the isotropic, B<sub>2</sub>, B<sub>4</sub> and crystalline phase sequence. Atomic force microscopy (AFM) was used to characterize the morphological behaviour of the B<sub>4</sub> phase. The results are discussed with respect to results from known X-ray investigations of the same phase [3, 4].

## 2. Experimental

### 2.1. Dielectric results

As the sample, compound **1** was chosen [15]; this shows hysteresis with respect to the phase transitions. The data below were obtained by microscopic observation during cooling and heating with a rate of 5 K min<sup>-1</sup>.



**1**

Cr 397 B<sub>4</sub> 420 B<sub>2</sub> 426 (heating); I 424 B<sub>2</sub> 406 B<sub>4</sub> (cooling)

Dielectric studies were carried out in the frequency range from 0.1 Hz to 10 MHz using the Solartron Schlumberger Impedance Analyzer Si 1260 and a Chelsea Interface. A brass cell coated with gold ( $d = 0.05$  mm) was used as capacitor and was calibrated with cyclohexane. The sample could not be oriented, and first measurements in the frequency range from 1 Hz to 10 MHz were made during cooling. Dielectric absorption ( $\epsilon''$ ) and dispersion ( $\epsilon'$ ) curves of the two different phases are shown in figures 1 and 2.

\* Author for correspondence  
e-mail: kresse@chemie.uni-halle.de

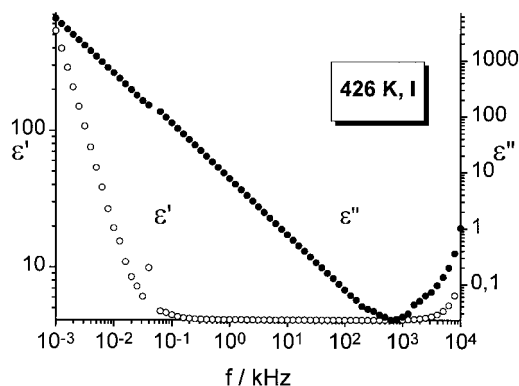


Figure 1. Dielectric constants ( $\epsilon'$ ) and losses ( $\epsilon''$ ) of **1** in the isotropic (I) phase.

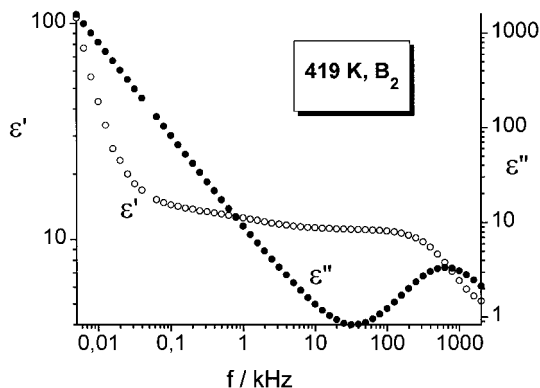


Figure 2. Dielectric constants ( $\epsilon'$ ) and losses ( $\epsilon''$ ) of **1** in the  $B_2$  phase.

The experimental data were fitted together to the real and imaginary parts of equation (1) to extract the specific behaviour of the different phases. The equation consists of two Cole–Cole mechanisms (terms 2 and 3), a conductivity contribution (term 4) and term 5 for the description of the capacitance of the double layer at low frequencies

$$\epsilon^* = \epsilon_2 + \frac{\epsilon_0 - \epsilon_1}{1 + (i\omega\tau_1)^{1-\alpha_1}} + \frac{\epsilon_1 - \epsilon_2}{1 + (i\omega\tau_2)^{1-\alpha_2}} + \frac{iD}{f} + \frac{E}{f^N} \quad (1)$$

where  $\epsilon_i$  are the low and high frequency limits of the dielectric constant,  $\omega = 2\pi f$  ( $f$  is frequency),  $\tau_i$  are the relaxation times, and  $\alpha_i$  are the Cole–Cole distribution parameters; the conductivity term, as well as  $E$  and  $N$  as further fit parameters describe the capacity of the double layer. The parameter  $D$  in the conductivity term allowed us to calculate the specific conductivity  $\sigma$  according to

$$\sigma = D2\pi\epsilon^0 \quad (2)$$

with  $\epsilon^0 = 8.85 \times 10^{-12} \text{ A s V}^{-1} \text{ m}^{-1}$ . For the low frequency mechanism, generally the limits of the dielectric

constants  $\epsilon_0$  and  $\epsilon_1$  are used. For the second, the limits  $\epsilon_1$  and  $\epsilon_2$  were taken. Due to this definition, the origins of the relaxation mechanisms in the different phases must not correspond to the limits of  $\epsilon_i$ .

The calculated limits of the dielectric constants are presented in figure 3. The  $B_2$  phase shows the expected increase in the dielectric constants. This is partly connected with the low frequency absorption related to a collective process, but arises mainly from the high frequency absorption connected with reorientation about the molecular long axes. The strong increase in intensity at the I– $B_2$  transition points to the formation of ferroelectric clusters in the short range [10, 11].

For the  $B_4$  phase, only the high frequency limit of the dielectric constants is given, as filled squares. It is important to point out that these data exceed the value of 3, but they are smaller than the high frequency limit in the  $B_4$  phase (open triangles). A dielectric absorption in the isotropic phase could not be found within experimental error. The measured dielectric constant of about 4 in the isotropic phase indicates that there must be at least one faster motion of the molecules which should be detected in the GHz range. This should be related to the reorientation of the molecules about their long axes [9]. Analysis of the data for the low frequency collective mechanism in the  $B_2$  phase results in distribution parameters of about  $\alpha_1 = 0.25$  for the low and  $\alpha_2 = 0.05$  for the high frequency process.

The specific conductivity calculated from the measured data in the range 1–100 Hz according to equation (2) is presented in figure 4. As in previous measurements [12], the conductivity decreases by about one decade at the transition into the  $B_4$  phase. This indicates a much higher degree of limitation of the dynamics in the  $B_4$  phase compared with the  $B_2$  state. It is important to note that the dielectric constant in the  $B_4$  phase remains constant for five days at room temperature and only after ten days was a small decrease in the dielectric constants detected. This gives a hint concerning the transition into the solid state.

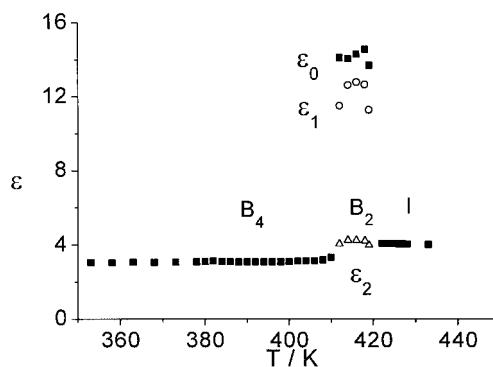
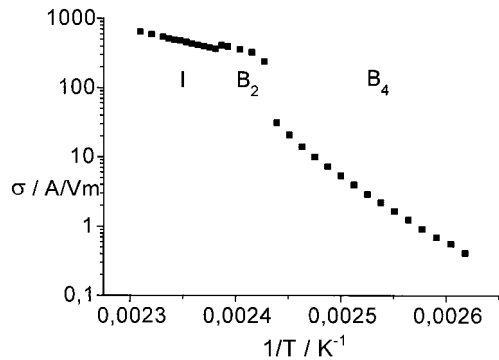
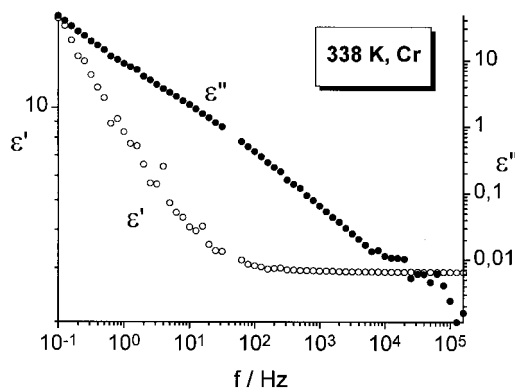
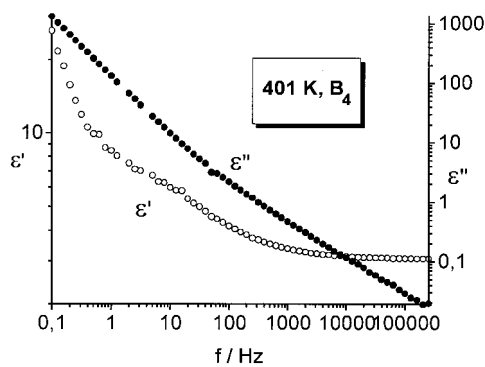


Figure 3. Limits of the dielectric constant for cooling


 Figure 4. Specific conductivity of **I**.

After the transition into the  $B_4$  phase, only a low frequency absorption was detected. It was difficult to extract good fit parameters from the data at low temperatures and therefore the frequency range was extended.

A second experimental run was performed during heating. At first the sample was heated to 383 K and stored for 5 hours at this temperature to transform the sample into the solid state. After that, dielectric measurements on heating were made. Dielectric data starting from 0.1 Hz in the crystalline state and after melting of the  $B_4$  phase are shown in figures 5 and 6. The main differences between figures 5 and 6 lie in the


 Figure 5. Dielectric data for **I** in the solid phase.

 Figure 6. Dispersion and absorption in the  $B_4$  phase.

relaxation range at 60 Hz and the higher dielectric constant in the  $B_4$  phase. The analysis of the data results in the high frequency dielectric constants which are presented in figure 7, together with the data from figure 3 obtained during cooling. It is obvious that at  $T = 397$  K the dielectric constant increases as a result of the melting process. Thus, we can conclude that in the  $B_4$  phase dynamics of dipoles are possible. This has to take place at higher frequencies ( $f > 10$  MHz) because the data given in figure 7 represent the high frequency limits of the dielectric constant in the  $B_4$  phase and no additional relaxation could be detected in the frequency range investigated.

The stronger fluctuations of the high frequency limits in the  $B_4$  phase result from the fitting process. The dielectric increments  $\Delta_1 = \epsilon_0 - \epsilon_1$  of the low frequency relaxation range and the relaxation times, respectively, are presented in figures 8 and 9. With consideration of the experimental errors, we can say that the low frequency process in the  $B_2$  and in the  $B_4$  phase do not differ strongly from each other. The same conclusion could be reached on the homologous sample in [15]. The data obtained from the cooling and heating processes also agree well. Furthermore, there is no indication

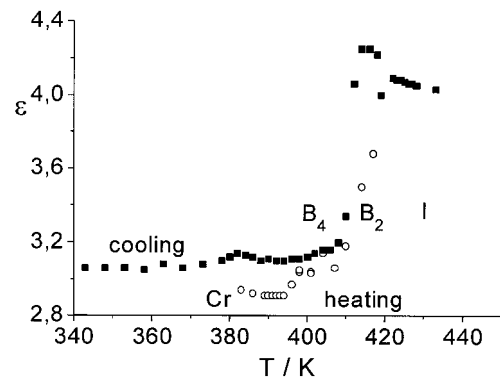


Figure 7. Calculated high frequency dielectric constants from data measured during heating and cooling.

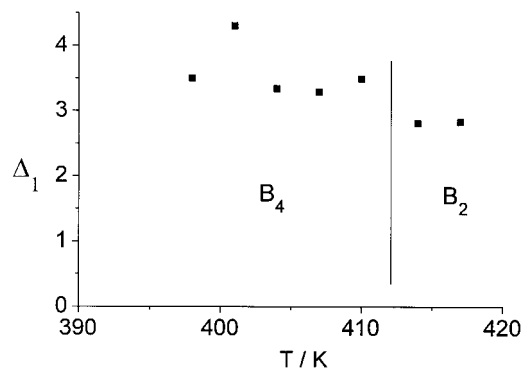


Figure 8. Dielectric increments calculated from data obtained by heating.

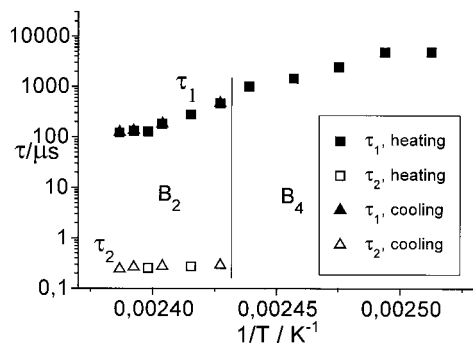


Figure 9. Relaxation times measured on heating and cooling

that the low frequency dynamics are influenced by the phase transition: neither a critical behaviour nor a marked step in the relaxation time is seen. This points to the fact that there is no ‘soft’ change in the superstructure like that of the soft mode at the phase transition SmC\*-SmA [7]. It has to be pointed out that the high frequency mechanism, interpreted as reorientation of the molecules about their long axes, could not be observed in the B<sub>4</sub> phase. Assuming a destruction of the ferroelectric short range order (decrease of intensity) and/or a much stronger hindrance of reorientation about the long axes of the molecules, it is possible that this relaxation

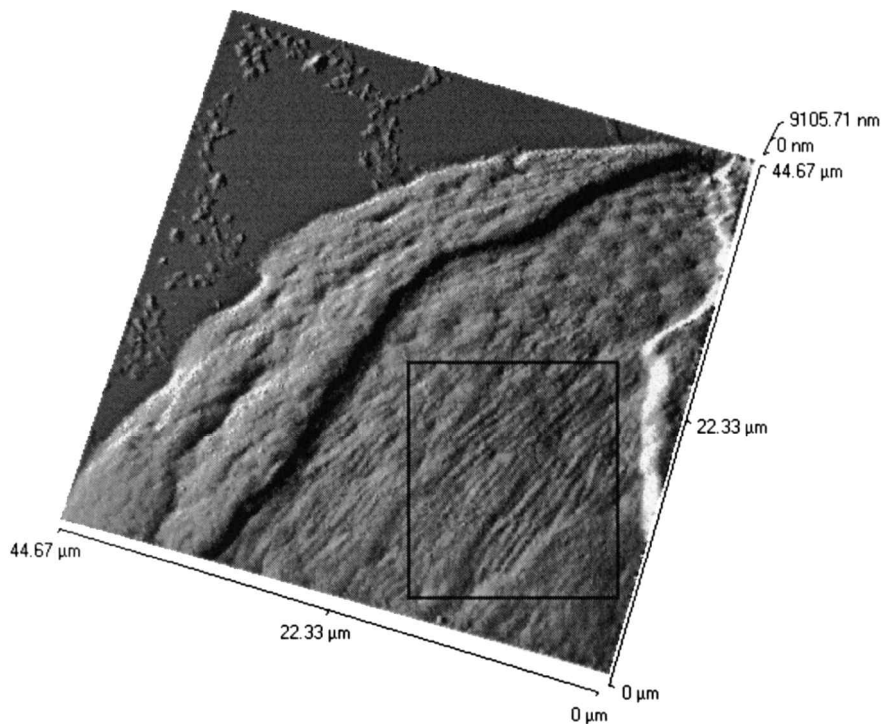


Figure 10. AFM image of a drop of the B<sub>4</sub> phase with focal-conic domains and periodic surface modulations (marked by a rectangle; see figure 11).

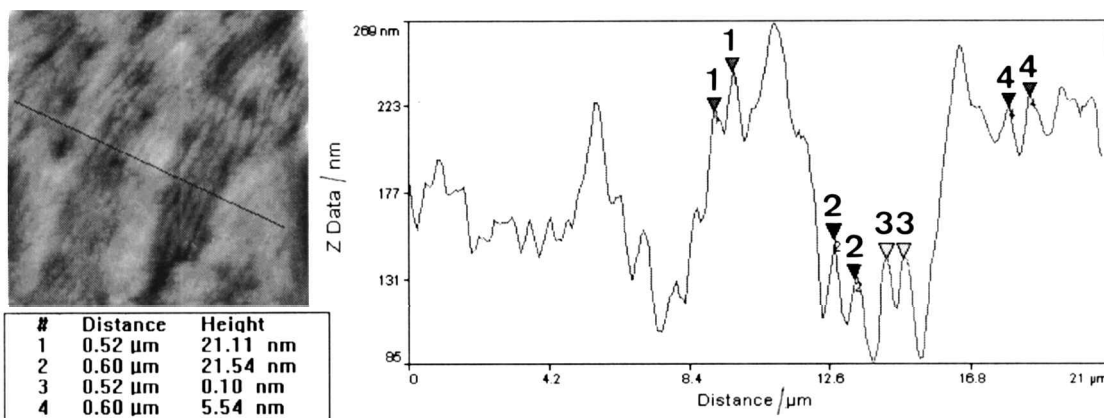


Figure 11. AFM image and line measurement of a detail from figure 10 (indicated by the rectangle) showing a periodicity of about 560nm. The data relate to the distance and height differences between points on the line indicated by the differently numbered pairs of inverted triangles.

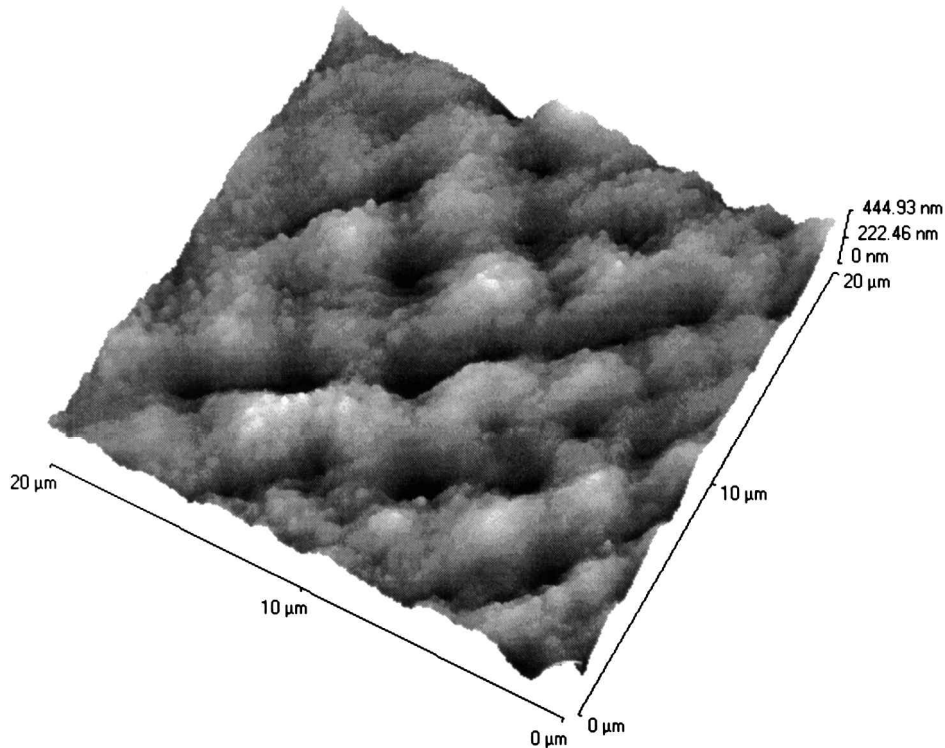


Figure 12. AFM image of the B<sub>4</sub> phase with lines of focal-conic domains at a distance of 4.4 μm and an additional disturbed periodicity of about 600 nm corresponding to that of figure 11.

cannot be detected experimentally. The last argument is supported by the fact that in the B<sub>4</sub> phase the sample becomes so firm that AFM measurements on thick samples could be performed.

2.2. AFM investigations

Atomic force microscopy is used to detect surface modulations caused by liquid crystalline superstructures. AFM measurements were performed at room temper-

ature and under ambient conditions using the TMX 2010 Discoverer of Topometrix (ThermoMicroscopes) in the non-contact mode with Si-tips (161 kHz resonance frequency). The maximal scan rate was 2 Hz. The image processing used was third order plane levelling.

The samples were prepared by filling a 10 μm sandwich cell of glass with compound 1, heating to the clearing temperature and cooling to room temperature. The cover plate was then removed and the samples were

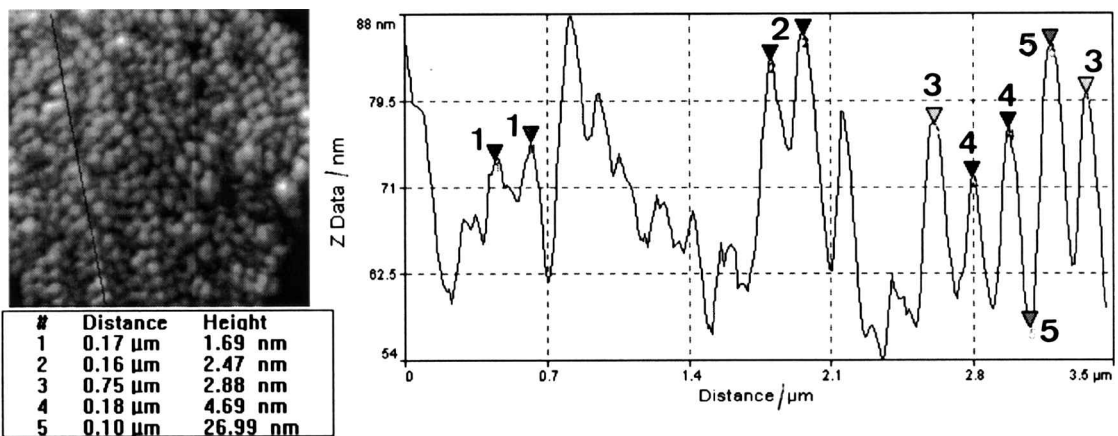


Figure 13. AFM image and line measurement of an area with a periodic surface pattern (period about 180 nm) for a sample of thickness 70 nm. The data are as described in the caption to figure 11.

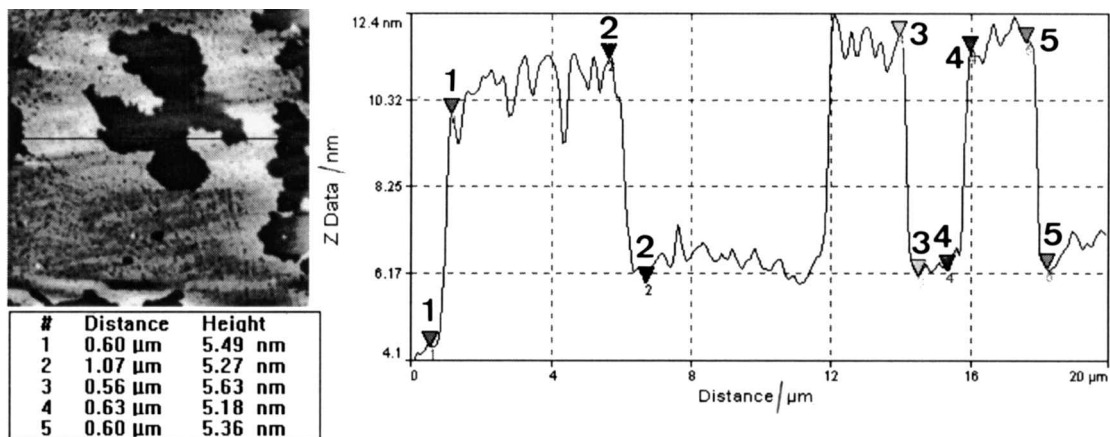


Figure 14. AFM image and line measurement of a monolayer area of the sample with a layer thickness of about 5.4 nm. The data are as described in the caption to figure 11.

again heated to the clearing temperature and cooled to room temperature using a cooling rate of  $0.5 \text{ K min}^{-1}$  in the neighbourhood of the clearing temperature, and then to the  $B_4$  phase at a cooling rate of  $5 \text{ K min}^{-1}$ . A sudden phase transition from  $B_2$  to  $B_4$  was observed by optical microscopy. The  $B_4$  phase was then rapidly cooled to room temperature.

Figure 10 shows the AFM image of a drop with a weak focal-conic surface texture and an additional superstructure periodicity visible in the area indicated by the rectangle.

Line measurements in the area of the superstructure reveal a periodicity of about 560 nm with a modulation of height of 40 nm (figure 11). It was found that in other

preparations there were regions of a few parallel  $0.6 \mu\text{m}$  lines where the direction of the lines changed from region to region. Thus, the origin of this periodicity cannot be related to the scanning process. On other regions of sample, this periodicity is partly suppressed by the focal-conic domains (figure 12). Such more pronounced focal-conic domains are shown in figure 12. This texture is typical for phases ordered in layers, e.g. liquid crystalline modifications [16]. Focal-conic domains were also observed by AFM in liquid crystalline  $B_2$  and  $B_7$  phases [17]. The funnel-shaped centres of the focal-conic domains are in keeping with a distance between the lines of  $4.4 \mu\text{m}$  (see figure 12). The difference in the height between the lines and the minimum of a funnel of the

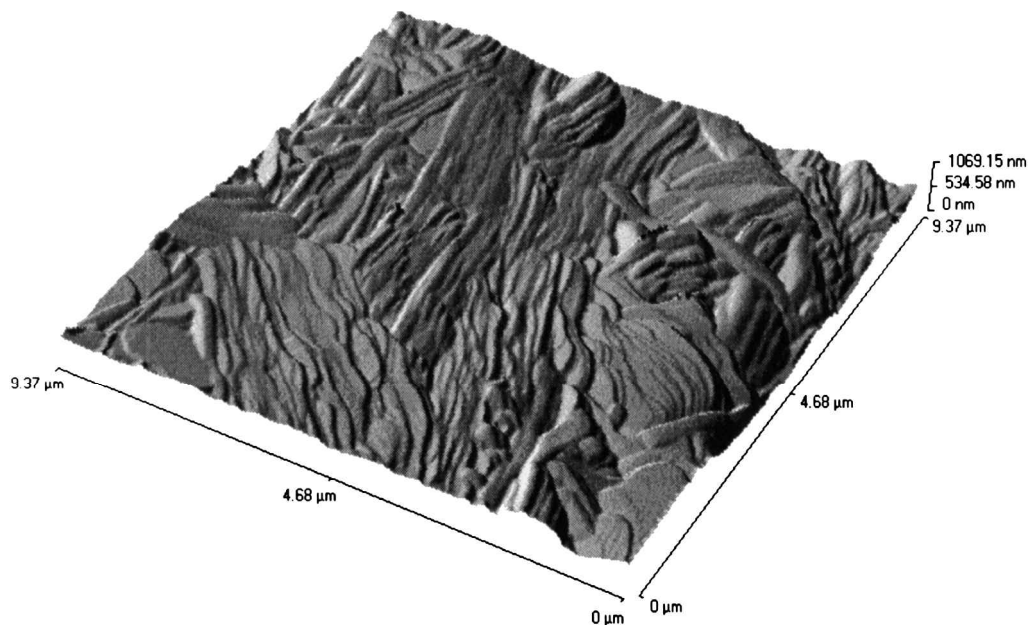


Figure 15. AFM image of the crystallized sample.

focal-conic domain was measured in some samples to be more than 1  $\mu\text{m}$ .

With higher magnification on sample areas with a thickness of about 70 nm an additional periodicity of about 180 nm in distance and 25 nm in height is recognisable (see figure 13). On a sample area with only a monolayer, a layer thickness of 5.4 nm was measured (figure 14). This distance corresponds to the length of the given molecule **1**. All the distances could be reproduced over different parts of the sample and even by changing the conditions for preparation.

After annealing the sample for 2 h at 100°C in the B<sub>4</sub> phase, the sample crystallized. An AFM image of the crystalline surface is given in figure 15. Focal-conic domains do not appear on the crystallized sample, for which totally different surface structures are found.

In conclusion one can say that at present the experimental data for the B<sub>4</sub> phase have to be discussed in contradictory terms. With respect to the X-ray pattern, which exhibits in the wide angle regions several diffuse bands [3, 4], the structure has to be discussed as solid-like, with the small particle size causing the broadening of the reflections. From the measurements presented here it can be concluded that the B<sub>4</sub> phase does not behave from a dynamical and morphological point of view like a classical crystal with a long range three-dimensional order and completely frozen dynamics. Thus, the B<sub>4</sub> phase seems to be a further example lying between high temperature liquid crystals and crystalline solids.

The authors are indebted to Dr Diele for discussions as well as the DFG and the Fonds der Chemischen Industrie for financial support.

## References

- [1] NIORI, T., SEKINE, J., WATANABE, J., FURUKAWA, T., and TAKEZOE, H., 1996, *J. mater. Chem.*, **6**, 231.
- [2] WATANABE, J., NIORI, T., SEKINE, J., and TAKEZOE, H., 1998, *Jpn. J. appl. Phys.*, **37**, L 139.
- [3] DIELE, S., PELZL, G., and WEISSFLOG, W., 1999, *Liq. Cryst. Today*, **9**, 8.
- [4] PELZL, G., DIELE, S., and WEISSFLOG, W., 1999, *Adv. Mater.*, **11**, 707.
- [5] LINK, D. R., NATALE, G., SHAO, R., MCLENNAN, J. E., CLARK, N. A., KÖRBLOVA, E., and WALBA, D. M., 1997, *Science*, **278**, 1924.
- [6] BRAND, H. R., CLADIS, P., and PLEINER, H., 1998, *Eur. Phys. J. B*, **6**, 347.
- [7] BOURNEY, V., PAVEL, J., LORMAN, V., and NGUYEN, H. T., 2000, *Liq. Cryst.*, **27**, 559.
- [8] SCHMALFUSS, H., SHEN, D., TSCHERSKE, C., and KRESSE, H., 2000, *Liq. Cryst.*, **27**, 1235.
- [9] WEISSFLOG, W., LISCHKA, CH., DIELE, S., PELZL, G., WIRTH, I., GRANDE, S., KRESSE, H., SCHMALFUSS, H., HARTUNG, H., and STETTLER, A., 1999, *Mol. Cryst. liq. Cryst.*, **333**, 203.
- [10] SCHMALFUSS, H., SHEN, D., TSCHERSKE, C., and KRESSE, H., 1999, *Liq. Cryst.*, **26**, 1767.
- [11] SCHMALFUSS, H., WEISSFLOG, W., HAUSER, A., and KRESSE, H., 2000, *Proc. SPIE*, **4147**, 172.
- [12] SALFETNIKOVA, J., SCHMALFUSS, H., NADASI, H., WEISSFLOG, W., and KRESSE, H., 2000, *Liq. Cryst.*, **27**, 1663.
- [13] SCHMALFUSS, H., HAUSER, A., and KRESSE, H., *Mol. Cryst. liq. Cryst.* (in the press).
- [14] SCHMALFUSS, H., WEISSFLOG, W., HAUSER, A., and KRESSE, H., *Mol. Cryst. liq. Cryst.* (in the press).
- [15] NADASI, H., LISCHKA, CH., KRESSE, H., WIRTH, I., DIELE, S., PELZL, G., and WEISSFLOG, W., submitted.
- [16] BOULIGAND, Y., 1998, *Handbook of Liquid Crystals*, Vol. 1, edited by D. Demus, J. Goodby, G. W. Gray, H.-W. Spiess and V. Vill (Weinheim: Wiley-VCH), pp. 406–453.
- [17] HAUSER, A., SCHMALFUSS, H., and KRESSE, H., 2000, *Liq. Cryst.*, **27**, 629.